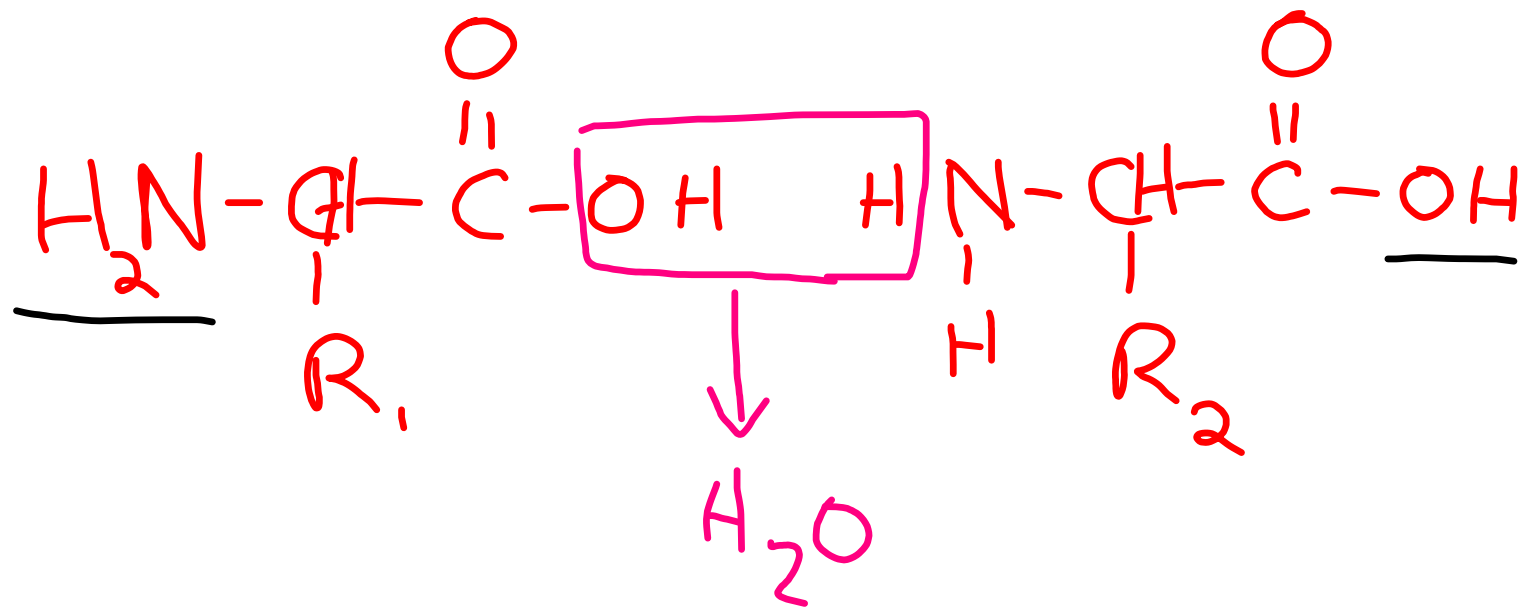


amino acid

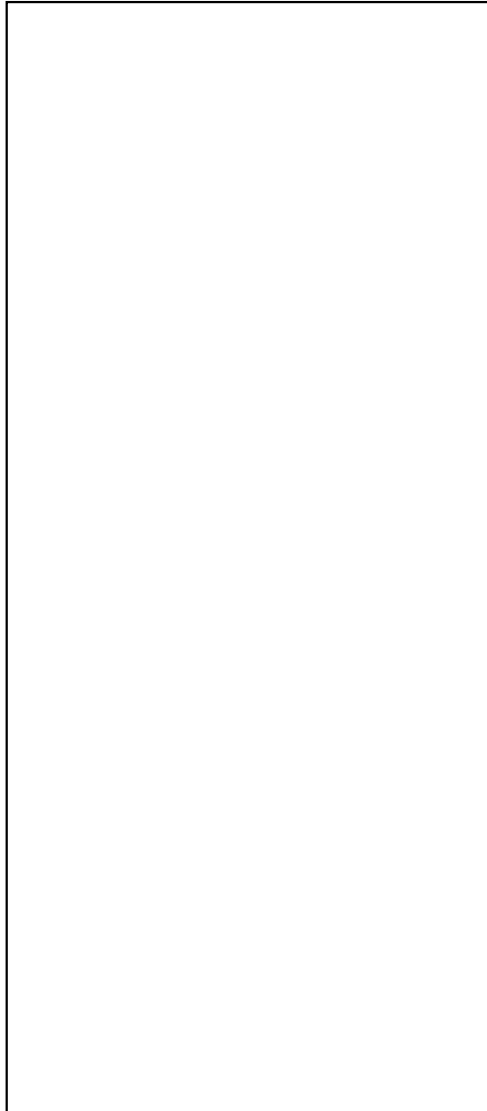
$\text{R}_1$  = alkyl chain  
or a hydrogen

A peptide bond is formed via covalent binding of the carbon atom of the Carboxy group of one amino acid to the nitrogen atom of the amino group of another amino acid by dehydration.



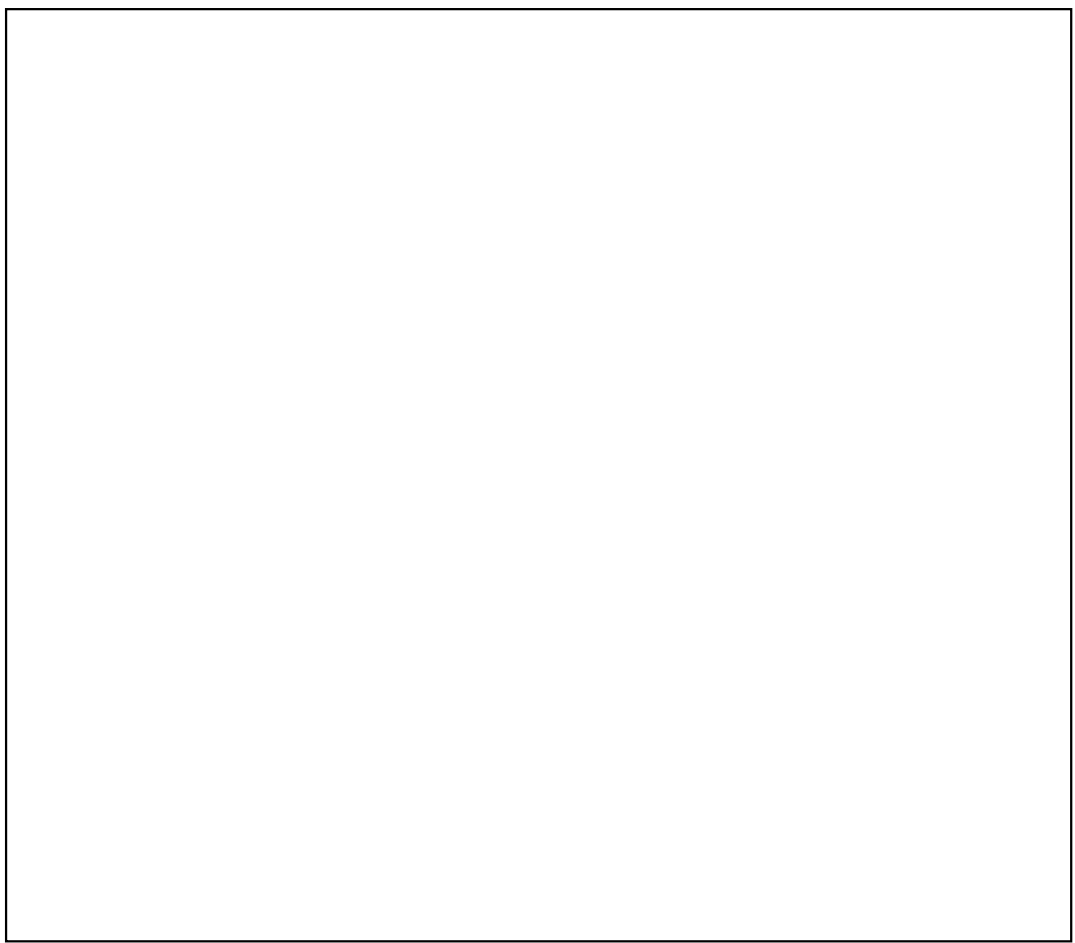


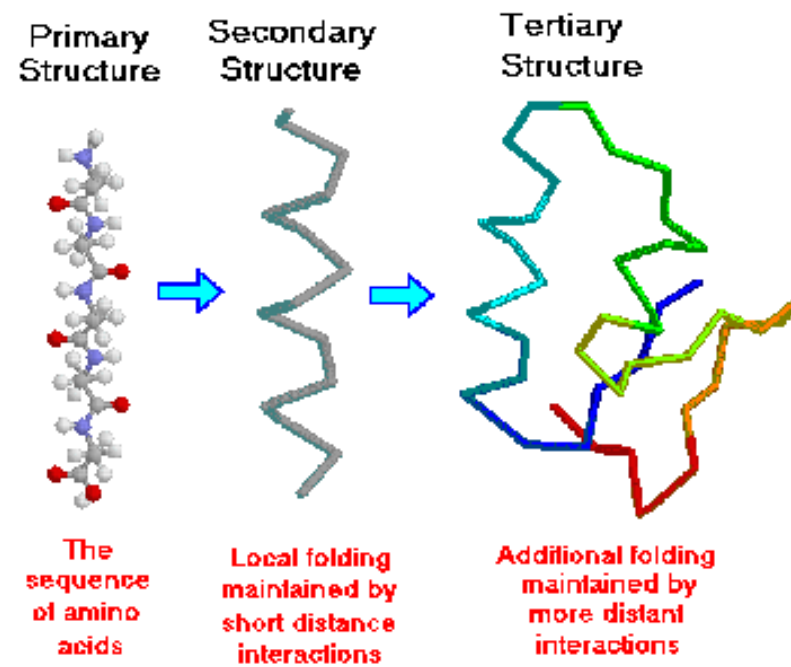
Secondary Structure is a local regulary occurring structure in proteins and is mainly formed through hydrogen bonds between backbone atoms. So-called random coils, loops or turns don't have a stable secondary structure. There are two types of stable secondary structures: **Alpha-helices and beta-sheets**. Alpha-helices and beta-sheets are preferably located at the core of the protein whereas loops prefer to reside in outer regions



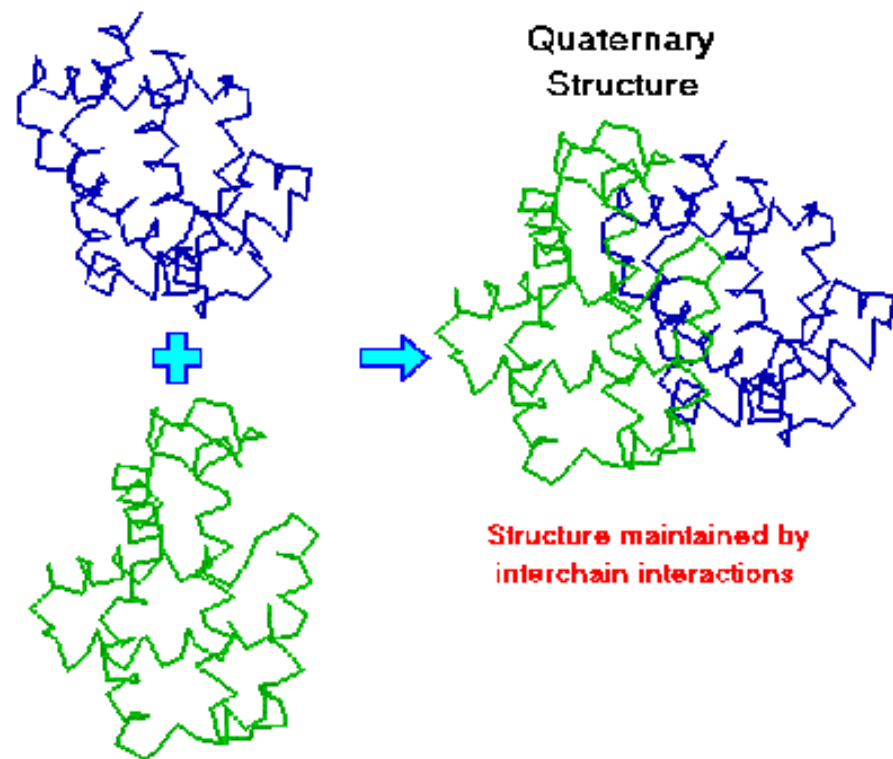
N  
|  
H

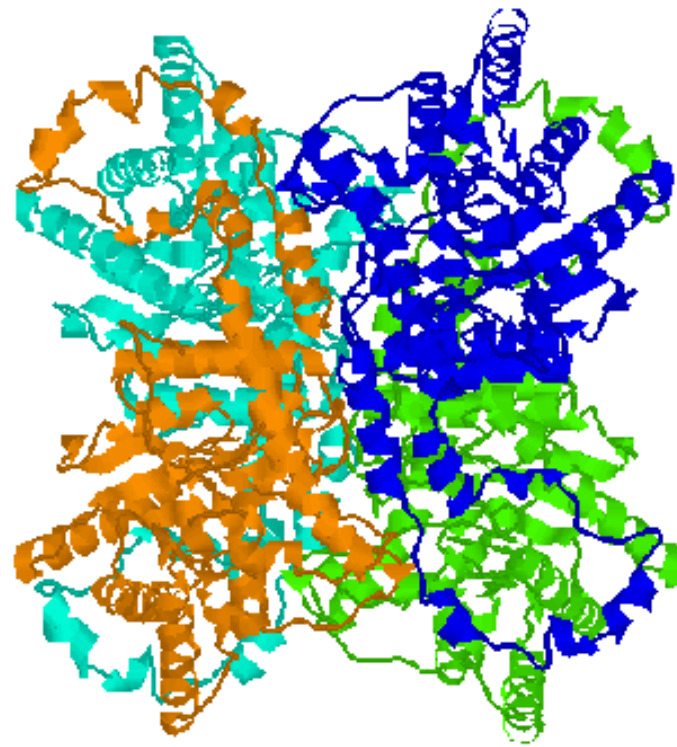
Tertiary structure describes the packing of the alpha-helices, beta-sheets and random coils with respect to each other on the level of one whole polypeptide chain.

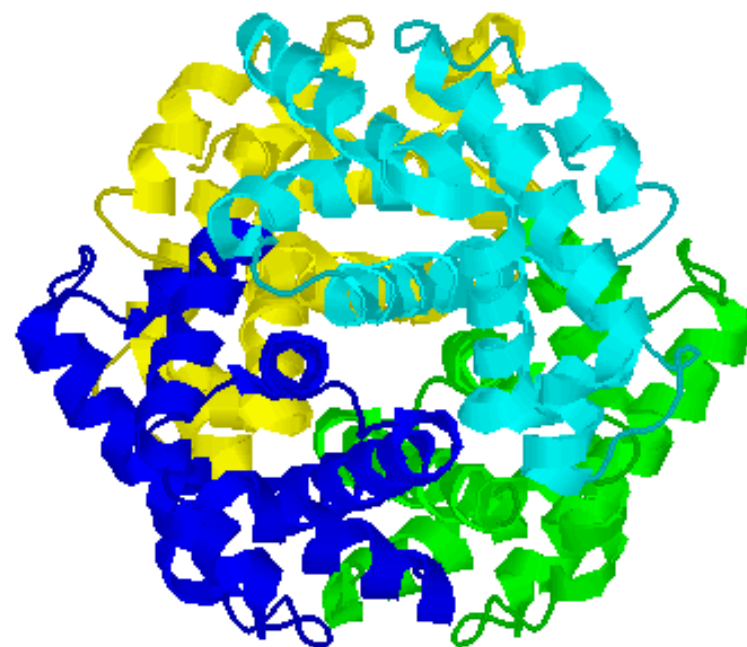




Quaternary Structure -  
the combination of two  
or more chains, to form  
a complete unit. The  
interactions between  
chains are not different  
from those in tertiary  
structures, but are  
distinguished only by  
being interchain rather  
than intrachain.







Radiation Lab 10<sup>th</sup>

Distillation Lab 17<sup>th</sup>

Test III 24<sup>th</sup>

Lab-Lab

May 1<sup>st</sup>

May 8<sup>th</sup>

1:30 - 4:00

Tues.

6. N, P, A, B

7. Put together an amino acid chain

12.

Isomer - same  
molecular formula, but  
differently built.

Stereoisomer-  
mirror image  
Carbon atom must  
have 4 different  
attachments to it.

